Preliminary Studies on the Biosynthesis of the Terpenoids in Artemisia monogyna

By

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The present work was carried out as a preliminary investigation on the biosynthesis¹⁾ of santonin and the essential oils by *Artemisia monogyna*.

(2-C¹⁴) Mevalonic acid²⁾⁻⁵⁾ and 5-phosphomevalonic acid⁶⁾⁷⁾ were used as precursors of the terpenoid biosynthesis in the plant. Terpenoids were separated into several fractions by using various solvents and radioactivity incorporated into each fraction was determined by using a gas flow counter.

Materials and Methods

Labeled compounds and other chemicals.

The DL-[2-C¹⁴] mevalonic acid used was obtained from the Radiochemical Centre, Amersham (England). Radioactive L-5-phosphomevalonic acid was prepared from DL-[2-C¹⁴] mevalonic acid by enzymic preparation of *Staphyllococcus aureus*⁸).

Petroleum benzin used as the solvent was refined by washing with concentrated H_2SO_4 , mixed solution of concentrated H_2SO_4 and concentrated HNO_3 , 2% solution of NaOH, and water in succession. α -Santonin used as a carrier was supplied by the Nippon Shinyaku Company Ltd. Veronalacetate, β -indoleacetate, and other reagents were commercial goods.

Administration of labeled compounds.

Artemisia monogyna (Mibuyomogi) used in this experiment were Yamashina No. 3 and a common species.

The plants were cut to the approximately equal sizes (about 20 cm. length), and were dipped into an aqueous solution containing a labeled compound and other chemicals as shown in Table 1. After a few days cultivation, the plants were harvested and were dealed with by the following chemical treatment. The details of dip method carried out in this study are given in Table 2.

Separation of radioactive products.

The plants were cut to pieces and ground with a mortar. To this homogenate 25mg, of α -santonin was added as a carrier, and 1 ml. of ethanol was also added to stop enzymatic reactions. After boiling for 10 minutes under reflux the mixture was extracted with petroleum benzin, chloroform, and distilled water. The each extract was washed, dried, and the radioactivity was measured as infinitely thin samples in a 2π gas flow counter. From the chloroform solution crude santonin was obtained by the usual chemical method as calcium santoninate. The essential oil was separated into a petroleum benzin fraction.

Results and Discussion

The results obtained using DL-(2-C14) mevalonic acid and L-(2-C14) phosphomevalonic acid are shown in Table 2.

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Table 1

Exp.	No.	Labeled compound (×10 ³ c.p.m.)		Added chemicals (µ moles)					
		DL-MVA	L-P-MVA	Mn ⁺⁺	ATP	Phosphate buffer	Veronal acetate	β-Indole acetate	
I	1 2 3 4	20 20 20 20 20		2 2 —	5 - 5 -	10 10 10 10	=		
11	1 2 3 4	20 20 20 20 20		2 2 —	5 — 5 —	10 10 10 10		_ _ _	
Ш	1 2 3	20 20 20		2 2 2	_	10 10 10			
1V	1 2 3 4 5 6	40 40 40 40 40			 5 	10 	10 10 10 10	——————————————————————————————————————	
V	1	200		_	_				
VI	1	_	20				_		
VII	1 2 3 4	200 200 100 100		10			= =	0.01 1.00	
VIII	1	100			_		_		

The incorporation of C¹⁴ into santonin and essential oil fractions were observed. The rate of incorporation into these fractions were varied with the growth phase of the plant.

For incorporation into santonin, administration of a large amount of mevalonic acid and phosphomevalonic acid were advantageous. The effects of manganese ion and ATP were not so distinct under these cultivation. While, in the dark the incorporation of mevalonic acid into santonin was scarcely observed. This fact suggests that the biosynthesis of santonin may be occurred joining with photosynthesis by uncertain ways.

The plants used in this study were kindly supplied by the Nippon Shinyaku Institute for Botanical Research.

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Table 2

Dip method				C ¹⁴ incorporation (c.p.m.) Fraction				1	
Exp.	Date	Species	Weather and Lighting	No.	Petroleum benzin	Chloroform	Water	Total	Crude santonin
Ι	Jul. 12	common	fine	1 2 3 4	1,900 600 300 900	700 500 500 700	2,400 500 500 400	5,000 1,600 1,300 2,000	
П	Jul. 14	common	fine	1 2 3 4	300 2,300 1,600 2,200	2,000 3,200 2,100 1,600	6,000 6,600 4,700 3,100	8,300 12,100 8,400 6,900	250 10 225 200
III	Jul. 22	Yamashina No. 3	fine	1 2 3	400 100 300	2,400 2,100 1,900	4,000 6,460 2,260	6,800 8,660 4,460	200 225 300
1V	Jul. 25	Yamashina No. 3	cloudy	1 2 3 4 5 6	5,300 300 700 1,100 200 300	8,100 3,400 4,400 4,100 2,200 600	15,700 8,300 12,350 9,600 3,900 500	29,100 12,000 17,450 14,800 6,300 1,400	275 350 50 50 100 250
V	Aug. 5	Yamashina No. 3	fine	1	23,350	16,800	64,200	104,350	1,700
VI	Aug. 14	common	a fluorescent light	1	5,110	2,710	2,720	10,540	420
VII	Aug. 18	Yamashina No. 3	a fluorescent light	1 2 3 4	20,450 20,850 12,100 8,150	13,100 15,550 7,900 4,900	48,800 57,900 32,350 28,950	82,350 94,300 52,350 42,000	216 110 10 140
VII	Sep. 12	common	in the dark	1	24,500	5,600	35,625	65,725	0